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underneath each reaction description,  
and answer the question. (a) A solution

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of ammonia is added to a dilute solution of acetic acid. Identify the conjugate acid-base pairs in this reaction.

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(II) ions. b) Write a balanced chemical equation for the reaction of aluminum with one of the compounds you suggested in a). c) ... Write molecular, ionic and net ionic equations for any reactions that occur between the following pairs of compounds.

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example, often remains either mostly or completely relevant to the new LO's that deal with  $K_{sp}$ ; 2.

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chemical reaction at least one new substance must be produced. Recall that there are definite signs that a new substance gets produced a) a change in colour b) the formation of a gas c) the formation of a precipitate

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